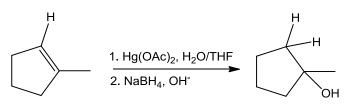
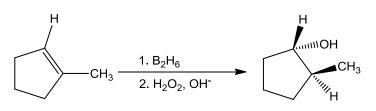
Alcohols

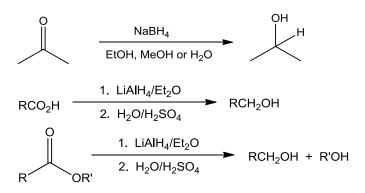
1. Oxymercuration



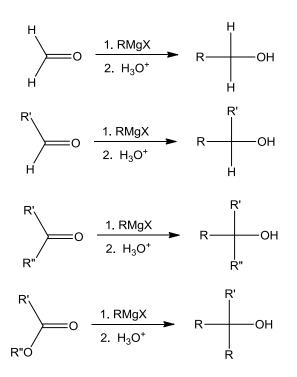
2. Hydroboration



3. Reduction of aldehydes, carboxylic acids, esters, and ketones

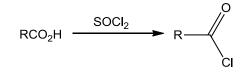


4. From Grignard reagents



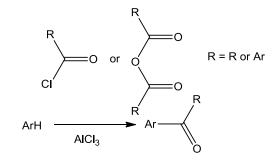
Acyl chlorides

1. From carboxylic acids

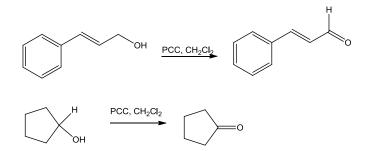


Aldehydes and ketones

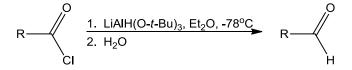
1. Friedel – Crafts acylation



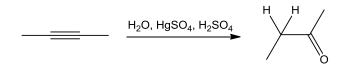
2. Oxidation of alcohols



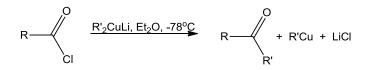
3. Reduction of acyl chlorides



4. From alkynes

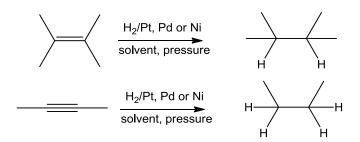


5. From lithium dialkylcuprates



Alkanes

1. Hydrogenation of alkenes and alkynes



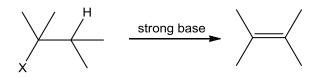
2. Reduction of alkyl halides

3. Corey - Posner, Whitesides - House synthesis

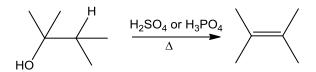
RX
$$\xrightarrow{1. \text{ Li}}$$
 R₂CuLi
R₂CuLi $\xrightarrow{R'X (1^{\circ}) \text{ or } ArX}$ R-R ou R-Ar

Alkenes

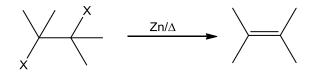
1. Dehydrohalogénation of alkyl halides



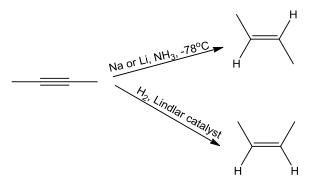
2. Dehydration of alcohols



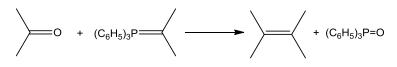
3. Dehalogenation of vicinal dihalides



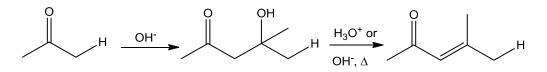
4. Hydrogenation of alkynes



5. Wittig reaction

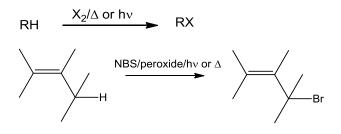


6. Aldol condensation

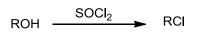


Alkyl halides

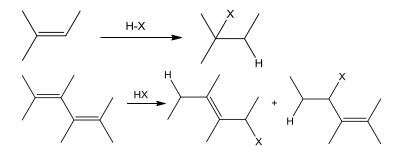
1. Halogenation of alkanes



2. From alcohols



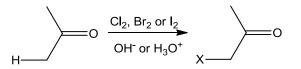
3. From alkenes



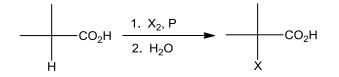
4. Finkelstein reaction

RX Nal/acetone

5. Halogenation of aldehydes and ketones

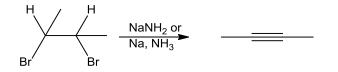


6. Hell-Volhard-Zelinski reaction



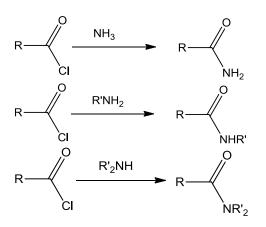
Alkynes

1. From vicinal dihalides

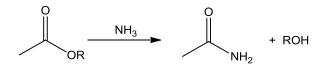


Amides

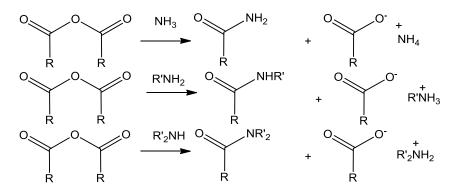
1. Ammonolyse des chlorures d'acyle



2. Ammonolyse des esters

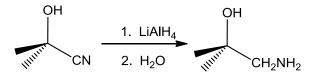


3. Ammonolyse des anhydrides



Amines

1. Reduction of nitriles



2. Hofmann degradation

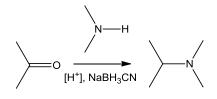
$$R \xrightarrow{O} \xrightarrow{OH^{-}} RNH_{2}$$

3. Alkylation of ammonia and amines

$$RX \longrightarrow RNH_2$$

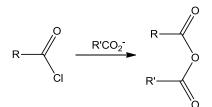
4. Reduction of -NO₂

5. Reductive amination



Anhydrides

1. From acyl chlorides



Aryl halides

1. Sandmeyer reaction

 $ArN_2^+ \xrightarrow{CuBr \text{ or } CuCl} ArBr \text{ or } ArCl$

2. Gattermann reaction

$$\begin{array}{c} Cu, HBr \text{ or} \\ Cu, HCl \\ ArN_2^+ & \longrightarrow & ArBr \text{ or } ArCl \end{array}$$

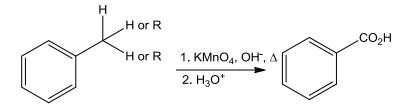
3. Synthesis of iodides

4. Synthesis of fluorides

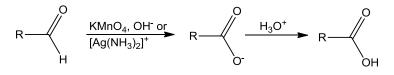
$$\operatorname{ArN}_{2^{+}} \xrightarrow{1. \operatorname{HBF}_{4}} \operatorname{ArF}$$

Carboxylic acids

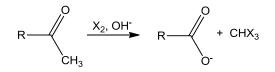
1. Oxidation of arenes



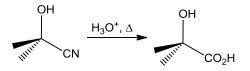
2. Oxidation of aldehydes



3. Haloform reaction



4. Hydrolysis of nitriles



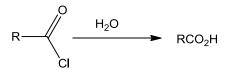
5. Oxidation of primary alcohols

$$\operatorname{RCH}_{2}\operatorname{OH} \xrightarrow{1. \operatorname{KMnO}_{4}, \operatorname{OH}^{-}, \Delta} \operatorname{RCO}_{2}\operatorname{H}_{3}\operatorname{O}^{+} \xrightarrow{} \operatorname{RCO}_{2}\operatorname{H}$$

6. From Grignard reagents

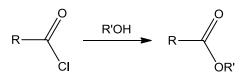
$$RMgX \xrightarrow{1. CO_2} RCO_2H$$

7. Hydrolysis of derivatives of carboxylic acids

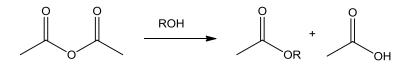


Esters

1. From acyl halides



2. From anhydrides

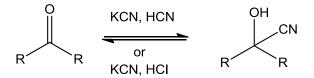


Nitriles

1. From primary alkyl halides

$$RCH_2X \xrightarrow{CN^-} RCH_2CN$$

2. From aldehydes and ketones



3. From diazonium salts