## Exercise 5 : - Acid-Base Reactions

 Camphor is terpene which has many varied uses such as a plasticizer and in fireworks. Even though it is a naturally occurring compound, the high demand for this compound requires an additional synthetic source. Camphene is a readily available terpene and is used in the synthesis. The first step in the reaction involves treatment of camphene with an acetic acid/sulfuric acid mixture.

Write an equation for the first step in this synthetic sequence.

2. The following table lists pKa values for a selection of acids in aqueous solution:

Acid	pKa	Acid	pKa	Acid	$pK_a$
HI	-10	CH <sub>3</sub> OH <sub>2</sub> +	-2,5	CH₃OH	16
H <sub>2</sub> SO <sub>4</sub>	-9	H₃O <sup>+</sup>	-1.74	(CH <sub>3</sub> ) <sub>2</sub> C=O	19.2
HBr	-9	CH <sub>3</sub> CO <sub>2</sub> H	4.75	HC≡CH	25
HCI	-7	NH <sub>4</sub> +	9.2	NH <sub>3</sub>	38
(CH <sub>3</sub> ) <sub>2</sub> OH <sup>+</sup>	-3.8	C <sub>6</sub> H <sub>5</sub> OH	9.9	H <sub>2</sub> C=CH <sub>2</sub>	44
(CH <sub>3</sub> ) <sub>2</sub> C=OH <sup>+</sup>	-2.9	CH <sub>3</sub> NH <sub>3</sub> <sup>+</sup>	10.6	C <sub>2</sub> H <sub>6</sub>	50

- a. Will dimethyl ether,  $(CH_3)_2O$ , dissolve in concentrated sulfuric acid,  $H_2SO_4$ ? Explain your answer.
- b. Will dimethyl ether, (CH<sub>3</sub>)<sub>2</sub>O, dissolve in acetic acid, CH<sub>3</sub>CO<sub>2</sub>H? Explain your answer.
- 3. Arrange the following compounds in order of decreasing acidity:

a. 
$$CH_3CH_2CH_3$$
  $H_3CC$   $CH_3CH=CH_2$ 

b.

$$\begin{array}{c|c} OH & OH \\ \hline O_2N & H_3C \\ \end{array}$$

c.

 $\mathsf{CH_3CH_2CO_2H} \qquad \mathsf{CH_3CHCICO_2H} \qquad \mathsf{CH_3CH_2CH_2OH}$